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(54) Title: MOISTURE CURABLE ALKOXYSLANE FUNCTIONAL POLY(ETHER-URETHANE) BASED SEALANTS

(57) Abstract

The present invention pertains to moisture curable sealant compositions based on alkoxy silane functional poly(ether-urethane)s which provide improved paint adhesion even in the absence of polar solvents or polar plasticizers. The present invention also provides coated substrates and laminates formed from such compositions.

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MOISTURE CURABLE ALKOXYSLANE FUNCTIONAL POLY(ETHER-URETHANE) BASED SEALANTS

5

FIELD OF THE INVENTION

10 The present invention pertains to moisture curable sealant compositions based on alkoxy silane functional poly(ether-urethane)s which provide improved paint adhesion even in the absence of polar solvents or polar plasticizers. The present invention also provides coated substrates and laminates formed from such compositions.

15

BACKGROUND OF THE INVENTION

20 Sealants are often used in automotive, marine, aerospace, and construction markets to fill gaps, to prevent corrosion, and to create aesthetically pleasing surfaces. Commercially viable sealants strike an acceptable balance among end use performance properties such as cure speed, shelf life, rheological characteristics, adhesion to various substrates, and thermal and ultraviolet light stability.

25 Consequently, such sealants typically contain rheology modifiers, adhesion promoters, oxidative stabilizers, plasticizers, and cure catalysts. A variety of technologies have been found useful in the manufacture of these sealants, including acrylic latexes, moisture curing silicones, and moisture curing isocyanate functional polymers. Moisture curing isocyanate functional polymers generally cure rapidly and, after cure, provide good adhesion to paint overcoats. Therefore, in applications which require that the sealant be painted soon after it is applied to a substrate, sealants based on moisture curing isocyanate functional polymer technology are particularly preferred. Even so, several problems attend the use of sealants based on moisture curing isocyanate functional polymers; under very hot, humid conditions these sealants are prone to foaming and the alcoholic co-solvents which are present in many paints react with the isocyanate termini of the moisture

curing isocyanate functional polymers thereby rendering them non-reactive and permanently preventing cure.

Recently, sealants based on moisture curing alkoxy silane functional polymers have been proposed as alternative to those based on moisture curing isocyanate functional polymers. The alkoxy silane functional polymers are preferred over isocyanate functional polymers especially because they do not foam under hot, humid conditions and they cure properly even in the presence of alcoholic paint co-solvents. It is possible, therefore, to paint sealants based on alkoxy silane functional polymers before they are fully cured (i.e., wet-on-wet) thereby eliminating the non-productive cure time required before painting sealants based on moisture curing isocyanate functional polymers.

Alkoxy silane functional polymers known in the art have been prepared by numerous methods including: condensation of isocyanate functional polymers with amine, mercaptan, or hydroxyl functional silanes; condensation of amine, mercaptan, or hydroxyl functional polymers with isocyanate functional silanes; coupling of mercaptan or hydrosilane functional polymers with alkene functional silanes; and coupling of alkene functional polymers with mercaptan functional silanes or alkoxyhydrosilanes. Alkoxy silane functional polymers may be compounded to form materials useful as moisture curable sealants. While a large number of alkoxy silane functional polymers has been disclosed in the art, none provide moisture curing sealants exhibiting acceptable paint adhesion when the paint is applied during the latter stages of cure.

U.S. Patent No. 3,627,722 (Seiter), U.S. Patent No. 4,067,844 (Barron et al.), U.S. Patent No. 4,857,623 (Emmerling et al.), and U.S. Patent No. 5,364,955 (Zwiener et al.) each disclose alkoxy silane functional poly(ether urethanes). None of these teaches a moisture curable sealant exhibiting acceptable paint adhesion when the paint is applied during the latter stages of cure.

U.S. Patent Nos. 5,476,889 and 5,464,888 (Owen) describe alkoxy silane functional polymer based sealant compositions in which the alkoxy silane endgroups are attached to a polypropylene oxide backbone through alkylene linking groups. While the initial paint adhesion of these sealant compositions was improved through

incorporation of polar plasticizers such as N-ethyl-*p*-toluenesulfonamide or polar solvents such as N-methylpyrrolidinone, the paint adhesion decreases to unacceptable levels as the alkoxysilane functional polymer cures.

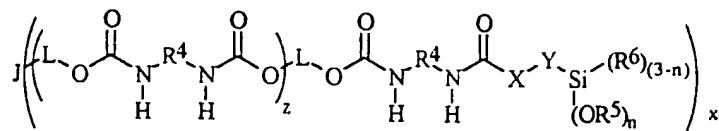
5 **SUMMARY OF THE INVENTION**

In spite of the considerable advantages provided by sealants based on alkoxysilane functional polymers, we have observed an unacceptable, time dependent decrease in the adhesion of paint to the previously known moisture curable sealants based on alkoxysilane functional polymers. In fact, we have 10 observed that previously known alkoxysilane functional polymers provide only minimal paint adhesion when the sealants are allowed to cure completely prior to paint application. This exactly opposes what is observed with sealants based on moisture curing isocyanate functional polymers. Incorporation of additives such as 15 polar plasticizers and polar solvents into sealants based on alkoxysilane functional polymers improves the adhesion of paint during initial stages of cure, but the improvement is insufficient for many applications and the presence of polar solvents and polar plasticizers is often problematic due to migration and high cost.

Therefore, there exists an unmet need for moisture curing sealants based on alkoxysilane functional polymers which provide improved paint adhesion even in the 20 absence of polar solvents or polar plasticizers.

The present invention pertains to moisture curable sealant compositions based on alkoxysilane functional poly(ether-urethane)s which provide improved paint adhesion even in the absence of polar solvents or polar plasticizers. The present invention also provides coated substrates and laminates formed from such 25 composition. The moisture curable sealant composition of the present invention comprises an alkoxysilane functional poly(ether-urethane) having formula (I):

(I)



wherein:

J is selected from the group consisting of alkyl, aryl, alkylene, alkarylene, and arylene with valence x and a number average molecular weight of less than about 250, preferably methyl, ethyl, methylene, ethylene, or 1,1,1-propanetrimethylene;

5 each L is independently selected from the group consisting of (i) a polyether segment comprising greater than 15 mol% and less than 40 mol% ethylene oxide (EO) units distributed randomly or in blocks (the segment preferably comprising 20 to 30 mol% EO) and the segment having a number average molecular weight of about 2000 to about 8000 (preferably about 3000 to about 6000), and (ii) a divalent alkylene group having a number average molecular weight of less than about 200;

10 each R⁴ is independently selected from the group consisting of C₂ to C₁₀ alkylene and cycloalkylene, preferably 1,6-hexylene and 1-(3-methylene)-3,5,5-trimethylcyclohexenyl, most preferably 1-(3-methylene)-3,5,5-trimethylcyclohexenyl (i.e., isophorone-diyl);

15 each z is independently selected from the group consisting of integers greater than or equal to zero, preferably 0 to 4;

 each X is independently selected from the group consisting of O, S, and NR⁷, wherein R⁷ is selected from the group consisting of hydrogen, aryl, alkyl, -CH₂CH(R)W¹ and -CH(W²)CH₂W¹, wherein

20 each R is independently selected from the group consisting of C₁ to C₄ alkyl; and

 W¹ and W² are independently selected from the group consisting of -CO₂R⁹, -CONR⁹₂, -CN, and -COR⁹; or W¹ and W² taken together form a divalent linking group selected from the group consisting of -CO₂-R⁸-O₂C-,

25 -CONR⁹-R⁸-NR⁹CO-, -CONR⁹-R⁸-O₂C-, and -CONR⁹CO- such that W¹, W², and an ethylene group through which they are joined form a 5 to 8 member ring, wherein R⁸ is independently selected from the group consisting of alkylene having from 2 to 10 carbon atoms and each R⁹ is independently selected from the group consisting of alkyl, and aryl;

each Y is independently selected from the group consisting of linear, branched, and cyclic alkylene groups having at least two (preferably 2 to 10) carbon atoms, most preferably -CH₂CH₂CH₂-;

each R⁵ is independently selected from the group consisting of alkyl, alkenyl, 5 aryl, acyl, and N-ketimanyl, preferably C₁ to C₄ alkyl;

each R⁶ is independently selected from the group consisting of hydrogen, alkyl, and aryl;

each n is independently selected from the group consisting of integers 1 to 3 (preferably 3); and

10 x is selected from the group consisting of integers greater than or equal to 1, preferably 1 to 6, most preferably 2 to 4.

In one embodiment each L is independently selected from the group consisting of (i) a polyether segment comprising an internal polypropylene oxide block and terminal polyethylene oxide blocks, the segment comprising 20 to 30 mol % EO and the segment having a number average molecular weight of about 3000 to 15 about 6000.

In one embodiment each L is independently selected from the group consisting of (i) a polyether segment comprising an internal polypropylene oxide block and terminal polyethylene oxide blocks, the segment comprising 20 to 30 mol % EO and the segment having a number average molecular weight of about 3000 to 20 about 6000;

each R⁴ is independently selected from the group consisting of 1,6-hexylene and isophorone-diyl.

each z is independently selected from integers 0 to 4;

25 each R⁵ is independently selected from the group consisting of C₁ to C₄ alkyl;

each n is the integer 3;

each x is independently selected from integers 2 to 4.

30 The moisture curable sealant composition of the present invention comprises a homogenous blend comprising:

- (a) 100 parts by weight of an alkoxy silane functional poly(ether-urethane) of formula I;
- 5 (b) 10 to 100 (more preferably 40 to 100) parts by weight of at least one plasticizer;
- (c) 0.1 to 10 (more preferably 3 to 8) parts by weight of at least one antioxidant;
- 10 (d) 1 to 5 (more preferably 1 to 3) parts by weight of at least one catalyst;
- (e) 0.1 to 10 (more preferably 3 to 6) parts by weight of at least one adhesion promoter;
- 15 (f) 0 to 10 (more preferably 1 to 5) parts by weight of at least one dehydrator;
- (g) 0 to 400 (more preferably 250 to 350) parts by weight of at least one filler;
- (h) 0 to 10 (more preferably 3 to 8) parts by weight of at least one rheology modifier; and
- 20 (i) 0 to 40 parts by weight of an organic solvent.

The present invention further comprises a laminate article, wherein the laminate comprises:

- 25 (a) a substrate;
- (b) a layer of the moisture curable sealant composition of the invention coated on at least a portion of the substrate; and
- (c) a layer of one or more of the following coated over the layer of the moisture curable sealant composition: (i) paint, (ii) primer, (iii) paint sealer.

30 The present invention even further comprises a cured laminate article,

wherein the laminate comprises:

- (a) a substrate;
- (b) a cured layer of the moisture curable sealant composition of the invention which is bonded to at least a portion of the substrate; and
- 35 (c) a layer of one or more of the following coated over the cured layer of the moisture curable sealant composition: (i) paint, (ii) primer, (iii) paint sealer.

A preferred embodiment of the present invention comprises a laminate article, wherein the laminate comprises:

- (a) a substrate;
- (b) a layer of the moisture curable sealant composition of the invention coated on at least a portion of the substrate;
- (c) a layer of primer coated over the layer of moisture curable sealant composition;
- 5 and
- (d) a layer of paint coated over the layer of primer.

A preferred embodiment of the present invention comprises a cured laminate article, wherein the laminate comprises:

- (a) a substrate;
- 10 (b) a cured layer of the moisture curable sealant composition of the invention bonded to at least a portion of the substrate;
- (c) a layer of primer coated over the cured layer of moisture curable sealant composition; and
- (d) a layer of paint coated over the layer of primer.

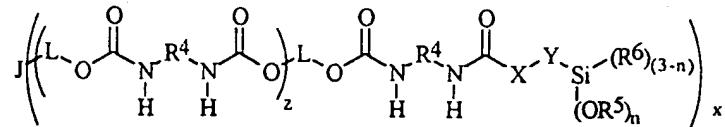
15 The laminate articles described above may include, for example, portions of land vehicles (such as automobiles, buses, trucks, vans, trains, etc.), aircraft, and watercraft (such as boats).

DETAILED DESCRIPTION OF THE INVENTION

A. Polymer of Formula (I) and its Formation

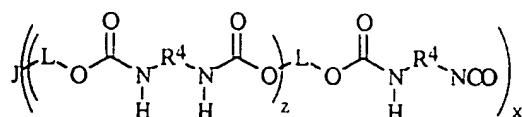
The present invention provides a moisture curable sealant composition comprising an alkoxy silane functional poly(ether-urethane) of formula (I) as previously defined:

25 (I)



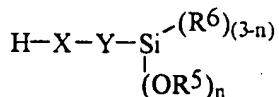
The alkoxy silane functional poly(ether-urethane) is the reaction product of an isocyanate functional poly(ether-urethane) of formula (IV):

(IV)



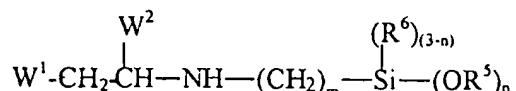
wherein J, L, R⁴, z and x are as previously defined,
and an endgroup precursor compound of formula (V):

5 (V)



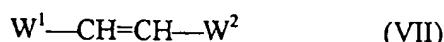
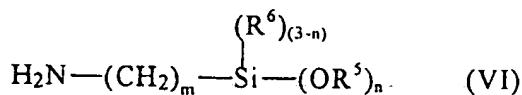
wherein R⁶, X, R⁵, Y and n are as previously defined.

10 A preferred endgroup precursor compound has the structure:



15 in which m is selected from the group consisting of integers 2 to 6 and R⁵, R⁶, W¹,
W², and n have the definitions provided above and which
may be formed, for example, by the Michael addition of a primary amine functional
alkoxysilane compound of formula (VI) with a Michael receptor of formula (VII) as
taught in U.S. Patent No. 5,364,955 incorporated by reference herein. Preferred
Michael receptors (VII) are maleic and/or fumaric acid diesters. Formulas (VI) and
(VII) are as follows:

20



in which R⁵, R⁶, W¹, W², m, and n have the definitions provided above.

Specific examples of preferred end group precursor compounds (V) include

25 HS(CH₂)₃(OCH₃)₃; H₂N(CH₂)₃Si(OCH₃)₃; H₂N(CH₂)₃Si(OCH₂CH₃)₃,
PhNH(CH₂)₃Si(OCH₃)₃; MeNH(CH₂)₃Si(OCH₃)₃;

BuO₂CCH₂CH₂NH(CH₂)₃Si(OCH₃)₃ and BuO₂CCH₂CH₂NH(CH₂)₃Si(OCH₂CH₃)₃
(disclosed in U.S. Patent No. 4,067,844); and
BuO₂CCH₂CH(CO₂Bu)NH(CH₂)₃Si(OCH₃)₃ and
BuO₂CCH₂CH(CO₂Bu)NH(CH₂)₃Si(OCH₂CH₃)₃ (disclosed in U.S. Patent No.
5,364,955).

A.i. Preparation of Isocyanate Functional Poly(ether-urethane)s

The isocyanate functional poly(ether-urethane) of formula (IV) may be prepared by the condensation reaction of one or more isocyanate reactive materials with an excess of one or more polyisocyanates, preferably diisocyanates, and preferably in the presence of a catalyst. Isocyanate reactive materials useful according to the invention include polyethers which have molecular weights of about 2000 to about 8000, EO contents greater than about 15 mol % and less than about 40 mol %, and one or more hydroxyl, mercaptan, primary or secondary amine groups. Isocyanate reactive materials useful according to the invention also include diols and diamines having a number average molecular weights of less than about 200 (known in the art as "chain extenders"). Blends of two or more isocyanate reactive materials may be used to make the isocyanate functional poly(ether-urethane)s. A summary of basic polyurethane chemistry can be found in *Polyurethanes: Chemistry and Technology*, Saunders and Frisch, Interscience Publishers (New York, 1963 (Part I) and 1964 (Part II)).

Preferred isocyanate reactive materials are block and random copolymers of ethylene oxide and propylene oxide having number average molecular weights in the range of from about 2000 to about 8000, more preferably from about 3000 to about 6000 and containing greater than about 15 mol% and less than about 40 mol% ethylene oxide (EO) units, more preferably about 20 mol% to about 30 mol%. Diols are especially preferred. When the EO content is less than or equal to about 15 mol%, little improvement in paint adhesion is observed. When the EO content is above about 40 mol %, the viscosity of the sealant often rises, moisture sensitivity increases, and compatibility with additives may be reduced.

Polyether diols are commercially available from a large number of sources and may be prepared by ring-opening polymerization reactions that are well known in the polymer art. The chemistry and mechanism of ring-opening polymerizations, for example, are discussed in detail in *Ring-Opening Polymerization* (Volumes 1, 2, and 3) edited by K.J. Ivin and T. Saegusa, 1984. Polyether diols useful as synthons in the present invention comprises ethylene oxide and higher alkylene oxide randomly or in blocks. Sequential polymerization of ethylene oxide (EO) and higher alkylene oxide, preferably propylene oxide (PO), will produce block copolymers, for example those having an internal polypropylene oxide block and terminal polyethylene oxide blocks, hereinafter denoted EO(PO)EO, and those having an internal polyethylene oxide block and terminal polypropylene oxide blocks, hereinafter denoted PO(EO)PO. Simultaneous polymerization of ethylene oxide and higher alkylene oxide, preferably propylene oxide, will produce random copolymers, hereinafter denoted EO/PO.

Numerous polyether diols are commercially available, for example from BASF Corporation, Mount Olive NJ; Dow Chemical Company, Midland MI; Olin Chemicals, Stamford CT; and ARCO Chemical Company, Newtown Square PA. A particularly preferred polyether diol is Pluronic L92 from ARCO, an EO(PO)EO block copolymer having 20 mol% EO units and number average molecular weight of about 3650.

Suitable polyisocyanates useful in preparation of isocyanate functional poly(ether-urethane)s of formula (IV) include aliphatic and cycloaliphatic polyisocyanates. Representative examples of useful polyisocyanates include the diisocyanates isophorone diisocyanate (hereinafter denoted IPDI) and 1,6-hexamediisocyanate. Dimers and trimers of the above mentioned diisocyanates, for example those containing uretdione, biuret, and isocyanurate linkages, are also contemplated. A preferred polyisocyanate is IPDI.

The condensation reaction to form an isocyanate functional poly(ether-urethane) is typically conducted in the presence of up to 5 % by weight (wt%) catalyst based on the isocyanate reactive material weight, preferably 0.02 to 0.4 wt%. Examples of useful catalysts include but are not limited to those listed in

Polyurethanes: Chemistry and Technology, Part I, Table 30, Chapter 4, Saunders and Frisch, Interscience Publishers, New York, 1963. Preferred catalysts are the tin IV compounds, for example dibutyltin dilaurate.

The reaction time required to convert the reactants to the desired isocyanate functional poly(ether-urethane)s will vary widely. Reaction times will depend upon several factors, including the nature of the reactants, the concentration of reactants, and the temperature of the reaction. Progress of the reaction is readily monitored by infrared (IR) spectroscopy by following the reduction of the isocyanate stretching frequency near 2270 cm^{-1} . Suitable reaction temperatures are usually between 0°C and 120°C , preferably between 25°C and 90°C , more preferably between 50°C and 90°C .

A.ii. Preparation of Alkoxy silane Functional Poly(ether-urethane)s

The alkoxy silane functional poly(ether-urethane) of formula (I) useful in the present invention is prepared by reacting an isocyanate functional poly(ether-urethane) of formula (IV) with the endgroup precursor of formula (V) described above. Preferably, the isocyanate functional poly(ether-urethane) has a number average molecular weight in the range of about 6000 to about 10,000, more preferably about 7000 to about 9000. Preferably, the ratio of equivalents of NCO to equivalents of endgroup precursor is 1:1. Preferably, a stepwise procedure is followed whereby the isocyanate functional poly(ether-urethane) is separately formed and then combined with the endgroup precursor to form the alkoxy silane functional poly(ether-urethane).

The reaction time required to convert the reactants to the desired alkoxy silane functional poly(ether-urethane)s of formula (I) will vary widely. Reaction times will depend upon several factors, including the nature of the reactants, the concentration of reactants, and the temperature of the reaction. Progress of the reaction is readily monitored by infrared (IR) spectroscopy by following the disappearance of the isocyanate stretching frequency near 2270 cm^{-1} . Suitable reaction temperatures are usually between 0°C and 120°C , preferably between 25°C and 90°C , more preferably between 50°C and 90°C .

B. Moisture Curable Sealant Compositions

5 The moisture curable sealant compositions of the present invention comprise at least one alkoxy silane functional poly(ether-urethane) of formula (I), at least one plasticizer, at least one antioxidant, at least one catalyst, and at least one adhesion promoter. The moisture curable sealant compositions of the present invention may further comprise one or more of the following optional ingredients: rheology modifiers, biocides, corrosion inhibitors, dehydrators, organic solvents, fillers, 10 colorants, photostabilizers, and perfumes. The components may be combined before or during the formation of the alkoxy silane functional poly(ether-urethane), provided that the components are not reactive with isocyanate. Preferably, the components are combined with the alkoxy silane functional poly(ether-urethane) after its formation. Each component is introduced to perform a specific function 15 and should be present at a level sufficient to perform that function.

20 Catalysts present in the moisture curable sealant compositions of the present invention enable the sealants to cure at rapid, controllable rates. Useful catalysts include, for example, metal salts and complexes, amines, and organic and inorganic acids. Specific examples of useful catalysts include dibutyltin diacetylacetone, tetraisopropyl titanate, calcium oxide, N,N,N',N'-tetramethylguanidine, tetrabutylammonium hydroxide, trifluoroacetic acid, and dibutyl phosphate.

25 Plasticizers present in the moisture curable sealant compositions of the present invention lower the viscosities of the sealants before cure and lower their moduli after cure. Useful plasticizers include, for example, benzoates, phthalates, adipates, and sulfonamides. It is generally preferred that the amount of plasticizer not exceed about 20% of the total weight of the composition. Specific examples of useful plasticizers include butyl benzyl phthalate, dipropylene glycol dibenzoate, dioctyl adipate, N-ethyl-*p*-toluenesulfonamide, diisodecyl phthalate, and dibutyl phthalate.

30 Adhesion promoters present in the moisture curable sealant compositions of the present invention provide good bonding between the sealants and the various substrates to which they are applied. Useful adhesion promoters include silanes, for

example, 3-(2-aminoethyl)aminopropyltrimethoxysilane, 3-glycidoxypropyltrimethoxysilane, and 3-mercaptopropyltrimethoxysilane.

Antioxidants present in the moisture curable sealant compositions of the present invention retard the oxidative degradation of its organic components. The presence of an antioxidant is especially important when the sealant is to be used in extreme temperature environments. Examples of useful antioxidants include but are not limited to those selected from the group consisting of hindered phenols and hindered amines. Specific examples of useful antioxidants include 2,6-di-*tert*-butyl-4-methylphenol, pentaerythritol tetrakis[3-(3,5-di-*tert*-butyl-4-hydroxyphenyl)propionate], bis(1,2,2,6,6-pentamethyl-4-piperidinyl) sebacate, and bis(2,2,6,6-tetramethyl-4-piperidyl) sebacate. Preferably, the antioxidant comprises a mixture of about 20 wt% hindered amine and 80 wt% hindered phenol.

Rheology modifiers which may be present in the moisture curable sealant compositions of the present invention cause the sealants to resist flow under static conditions, yet allow desirable flow while under shear. Useful rheology modifiers include, for example, castor waxes, fumed silicas, treated clays, and polyamides. Preferably, the rheology modifier is non-reactive with the alkoxy silane functional poly(ether-urethane).

Although not preferred because of environmental and energy considerations, one or more organic solvents may be present in the moisture curable sealant compositions of the present invention. Organic solvents reduce sealant viscosities. When used, preferred solvents are those which are unreactive with isocyanates and alkoxy silanes and include, for example, ketones, ethers, esters, amides, and hydrocarbons. Specific examples of preferred solvents include acetone, butanone, ethyl acetate, toluene, naphtha, N-methylpyrrolidinone, N,N-dimethylformamide, acetonitrile, tetrahydrofuran, 1-methoxy-2-propyl acetate, and Isopar H (an aliphatic hydrocarbon solvent available from Exxon). Especially preferred solvents are polar solvents. A specific example of an especially preferred solvent is N-methylpyrrolidinone.

Dehydrators which may be present in the moisture curable sealant compositions of the present invention react with any moisture which may enter the

sealant during processing or during storage and thus prevent premature gelation.

Specific examples of useful dehydrators include vinyltrimethoxysilane, trimethyl orthoformate, tetra(butanone oximino)silane, and vinyltri(butanone oximino)silane.

Fillers which may be present in the moisture curable sealant compositions of the present invention may be added to alter the color, rheology, and ultimate mechanical properties of the moisture curable sealant composition. Examples of useful fillers include carbon black, calcium carbonate, titanium dioxide, iron oxide, talc, ceramic microspheres, and clay. The fillers are preferably free of groups which react with the alkoxy silane functional poly(ether-urethane)s. Stearic acid surface treated, precipitated calcium carbonates are preferred fillers in applications where low cost and opacity are desirable.

Ultraviolet light stabilizers which may be present in the moisture curable sealant compositions of the present invention absorb ultraviolet radiation, thereby shielding the other organic components from this harmful energy. Useful ultraviolet light stabilizers include substituted hydroxyphenylbenzotriazoles. A specific example of a useful ultraviolet light stabilizer is 2-(3,5-di-*tert*-butyl-2-hydroxyphenyl)-5-chlorobenzotriazole.

C. Formulation of the Moisture Curable Sealant Compositions

The moisture curable sealant compositions of the invention may be formulated such that, they provide good adhesion to a variety of substrates, they provide good adhesion to a variety of paints, primers, and paint sealers, and they are sprayable, caulkable, or paste-like.

The physical properties of the moisture curable sealant compositions before and after cure will depend strongly on the amounts and identities of the components comprising the moisture curable sealant compositions. Viscosities of the moisture curable sealant compositions before cure will generally decrease, for example, with increasing levels of solvent or plasticizer, decreasing alkoxy silane silane functional poly(ether-urethane) molecular weight, decreasing filler concentration, and increasing filler particle size. A desirable viscosity for a sprayable moisture curable sealant composition, for example, would be between about 400 and about 700 Pa

sec (Brookfield spindle #7 at 2 rpm, 22 °C). Ultimate elongations of the moisture curable sealant compositions after cure will generally increase, for example, with increasing alkoxy silane silane functional poly(ether-urethane) molecular weight and plasticizer level.

5 Useful substrates for the articles of the invention include but are not limited to those selected from the group consisting of glass, metal, wood, and polymers. Useful metal substrates include, for example, cold rolled steel, primed steel, galvanized steel, and aluminum. Useful polymeric substrates include, for example, thermoplastics, paint coated surfaces, and fiberglass reinforced plastics.

10 Useful primers, paints, and paint sealers include for example those which utilize epoxy, acrylic, urethane, lacquer and enamel chemistries in commercially available formulations. Other paints, primers and paint sealers are also considered as useful and the above examples shall not be considered as limiting in any way.

15 D. Test Methods

Viscosity

Viscosities were determined at 22°C using a Brookfield DV-1+ viscometer and are reported in Pascal seconds (Pa sec).

20 Spectroscopy

IR spectra were obtained using a Nicolet 510 FT-IR spectrometer. NMR spectra were obtained using a Varion Unity 500 NMR Spectrometer. The NMR Spectrometer was operated at 500 megahertz to obtain ¹H-NMR spectra. All NMR runs were carried out using CDCl₃ solvent at 22°C using standard acquisition parameters.

25 Tack Free Time

This test was performed in a controlled environment having a temperature of 21°C and a relative humidity of 50%. A 0.64 cm bead of moisture curable sealant composition was applied to the test surface. Tack free time was the time required to produce a surface on the moisture curable sealant composition bead

which could be lightly touched by an applicator stick without transfer to the applicator stick.

Hardness

5 This test was performed in a controlled environment having a temperature of 21°C and a relative humidity of 50%. A 0.64 cm bead of moisture curable sealant composition was applied to the test surface. The hardness of the bead was measured after 24 hours (initial reading) and after seven days (final reading), using a Shore Durometer Type A.

10

Adhesion

A 0.64 cm diameter, 22.9 cm long bead of moisture curable sealant composition was applied to a cold rolled steel panel (30.5 cm x 10.16 cm) that had been cleaned by wiping first with methyl ethyl ketone, then with toluene, and then 15 again with methyl ethyl ketone. The bead was allowed to cure for one week in a controlled environment having a temperature of 21°C and a relative humidity of 50%. One end of the bead was cut away from the steel panel to form a free end. The free end was pulled, and the failure mode of the sealer was noted. Cohesive failure occurred when the sealer split, leaving sealer residue on the panel. Adhesive 20 failure occurred when the sealer lifted off the panel, leaving no residue. Of these two modes of failure, cohesive failure is preferred.

Cold Flexibility

25 A panel bearing a bead of cured, moisture curable sealant composition was prepared as described according to the adhesion test method above. The panel was cooled and held at -20°C for one hour. The panel was then quickly bent 180 degrees over a 2.54 cm diameter rod with the sealer on the outside radius. The sealer failed this test if it pulled away from the panel without leaving any residue (i.e., adhesive failure) or if it showed any cracks at the point of bending.

30

Wet-On-Wet Paintability

A 0.64 cm diameter, 22.9 cm long bead of moisture curable sealant composition was applied to a cold rolled steel panel (30.5 cm x 10.16 cm) that had been previously cleaned by wiping first with methyl ethyl ketone, then with toluene, and then again with methyl ethyl ketone. The bead was then smoothed to form a 0.16 cm thick film. Paint (PPG Deltron base clear available from Pittsburgh Paint & Glass, Inc. located in Strongsville, OH) was applied to separate films of the moisture curable sealant composition immediately and after the films of moisture curable sealant composition had aged for one hour, 24 hours, and 72 hours. The painting sequence was per the manufacturer instructions: One part base coat (DBU 9700) was mixed with 1.5 parts reducer (DRR 1170). Two applications of base coat were applied fifteen minutes apart using a spray pressure of 310 kilopascals. A minimum of twenty minutes later, two coatings of the clear coat (comprising 2 parts clear (DCU 2020), 1 part hardener (DU 5), and 1 part reducer (DT 870)) were applied to the base coat fifteen minutes apart using a spray pressure of 310 kilopascals. The paint surface was examined for the presence of cracking, wrinkling, or shrinkage. The resulting laminates comprising substrate, moisture curable sealant composition, and paint were allowed to age for three days. Each laminate was then examined to determine whether the paint and moisture curable sealant composition had cured properly. Proper paint cure was indicated by a dry paint surface. Proper moisture curable sealant composition cure was determined by cutting the sealer and examining whether it was dry throughout. If no defects were seen using these inspection procedures, the sealer was considered to have wet-on-wet paintability.

25

Paint Adhesion Test

A 0.64 cm diameter, 22.9 cm long bead of moisture curable sealant composition was applied to a cold rolled steel panel (30.5 cm x 10.16 cm) that had been previously cleaned by wiping first with methyl ethyl ketone, then with toluene, and then again with methyl ethyl ketone. The bead was then smoothed to form a 0.16 cm thick film. Paint (PPG Deltron base clear available from Pittsburgh Paint &

Glass, Inc. located in Strongsville, OH) was applied to separate films of the moisture curable sealant compositions immediately and after the films of moisture curable sealant composition had aged for one hour, 24 hours, and 72 hours. The painting sequence was per the manufacturer instructions: One part base coat (DBU 9700) was mixed with 1.5 parts reducer (DRR 1170). Two applications of base coat were applied fifteen minutes apart using a spray pressure of 310 kilopascals. A minimum of twenty minutes later, two coatings of the clear coat (comprising 2 parts clear (DCU 2020), 1 part hardener (DU 5), and 1 part reducer (DT 870)) were applied to the base coat fifteen minutes apart using a spray pressure of 310 kilopascals. The resulting laminates comprising substrate, moisture curable sealant composition, and paint were allowed to age for three days.

The paint adhesion of the moisture curable sealant compositions of the invention were then measured using ASTM 3359B. Thus, the test specimen was crosshatched with a razor, adhesive tape (3M Scotch™ 898 filament tape) was applied, and then withdrawn in a peel mode. The sample ratings corresponded to the percentage of paint film, as shown in the following chart:

Sample Rating	% Paint Removed
5B	0
4B	1-5
3B	6-15
2B	16-30
1B	31-60
0B	over 60

Preferably, a moisture curable sealant composition which has been painted one hour after its application to a substrate will have a paint adhesion value of at least 4B, more preferably 5B. Preferably, a moisture curable sealant composition which has been painted 24 hours after its application to a substrate will also have a paint adhesion value of at least 4B, more preferably 5B. Preferably, a moisture curable sealant composition which has been painted 72 hours after its application to a

substrate will have a paint adhesion value of at least 3B, more preferably at least 4B, and most preferably 5B.

Shelf Life

5 A cartridge containing the moisture curable sealant composition was tested for viscosity, tack free time, hardness, adhesion, and cold flexibility, and then stored at 50°C. After four weeks, the moisture curable sealant composition was re-submitted to the above tests. If the moisture curable sealant composition remained homogeneous and again passed these tests, then the moisture curable sealant
10 composition was considered to have good shelf life. Cartridges used for aging were high density polyethylene.

The following table defines the materials which were used in the examples.

15	Staflex DBM	Dibutyl maleate (C.P. Hall, Bedford Park, IL)
	Silquest A1100	3-aminopropyltriethoxysilane (OSi Specialties, Danbury, CT)
20	Silquest A1110	3-aminopropyltrimethoxysilane (OSi Specialties)
	Dabco T-12	dibutyltin dilaurate (Air Products, Allentown, PA)
25	Voranol 222-029	EO(PO)EO polyether diol, 20 mol% EO, 3900 MW (Dow Chemical, Midland, MI)
	Voranol 235-056	EO/PO polyether triol, 3500 MW (Dow Chemical)
30	Pluronic L62	EO(PO)EO polyether diol, 20 mol% EO, 2500 MW (BASF)
	Pluronic L92	EO(PO)EO polyether diol, 20 mol% EO, 3650 MW (BASF)
35	Pluronic 17R2	PO(EO)PO polyether diol, 20 mol% EO, 2150 MW (BASF)
	Multranol 9111	EO(PO)EO polyether diol, 4000 MW (Bayer, Inc.)
40	Pluronic L61	EO(PO)EO polyether diol, 10 mol% EO, 2000 MW (BASF)

	Pluronic L101	EO(PO)EO polyether diol, 10 mol% EO, 3800 MW (BASF)
5	Pluronic 31R1	PO(EO)PO polyether diol, 10 mol% EO, 3250 MW (BASF)
	Arcol E351	EO(PO)EO polyether diol, 15 mol% EO, 2800 MW (ARCO, Newtown Square, PA))
10	Desmodur I	isophorone diisocyanate (Bayer, Inc., Pittsburgh, PA)
	Mondur TD-80	Blend of 2,4- and 2,6-tolylene diisocyanates (80:20) (Bayer, Inc.)
15	Jayflex DIDP	diisodecyl phthalate (Exxon, Houston, TX)
	Tinuvin 770	bis(2,2,6,6-tetramethyl-4-piperidyl) sebacate (Ciba- Geigy, Hawthorne, NY)
20	Irganox 1010	pentaerythritol tetrakis[3-(3,5-di- <i>tert</i> -butyl-4- hydroxyphenyl)propionate] (Ciba-Geigy)
	Elftex 8	carbon black (Cabot, Inc., Boston, MA)
25	Dislon 6500	polyamide thickener (King Industries, Norwalk, CT) (currently named Disparlon 6500)
	Ultrapflex	0.07 micron, precipitated, stearated calcium carbonate (Specialty Minerals, Adams, MA)
30	Pfinyl 402	5.5 micron, ground, stearated calcium carbonate (Specialty Minerals, Adams, MA)
	Zeeospheres 600	ceramic microspheres (3M/Zeelan, St. Paul, MN))
35	Isopar H	aliphatic hydrocarbon solvent (Exxon)
	Silquest A171	vinyltrimethoxysilane (OSi Specialties)
40	Silquest A1120	3-(2-aminoethyl)aminopropyltrimethoxysilane (OSi Specialties)
	Neostann U220	dibutyltin diacetylacetone (Kaneka America, New York, NY)

The invention is further illustrated by the following examples, but the particular materials and amounts thereof recited in these examples, as well as other conditions and details, should not be construed to unduly limit this invention.

5

EXAMPLES

General Procedure for Formation of Secondary Amine Functional Alkoxy silane of Formula (V)

10 A mixture of Staflex DBM, Silquest A1100, and optionally Silquest A1110 (see Table 1) was prepared, exotherming spontaneously and raising the temperature to about 60°C. After 20 minutes, the infrared (IR) spectrum showed that the reaction had proceeded to about 90% completion. After 18 hours at room temperature, the proton nuclear magnetic resonance (¹H-NMR) spectrum indicated 15 about 97% conversion to the secondary amine functional alkoxy silane.

Table 1

Example	Staflex DBM	Silquest A1100	Silquest A1110
	wt (mmol)	wt (mmol)	wt (mmol)
1	45.6 gram (200)	22.1 gram (100)	17.9 gram (100)
2	57.0 gram (250)	55.25 gram (250)	0 (0)
3	45.6 gram (200)	11.5 gram (50)	26.85 gram (150)

General Procedure for the Preparation of Alkoxy silane Functional Poly(ether-urethane)

20 The polyether polyol was mixed with phosphoric acid (85 wt% solution in water, 0.14 mmol acid per kilogram polyol) and then dried over a four hour period at 105°C under pump vacuum (about 0.5 torr). The polyether polyol was allowed to cool to about 50°C in the absence of moisture and then mixed under a nitrogen 25 blanket with a diisocyanate and Dabco T-12. The mixture was agitated without

external temperature control for one half hour, then heated to and held at 70°C for three hours. The isocyanate functional poly(ether-urethane) thus obtained was allowed to cool to about 50°C, a secondary amine functional alkoxy silane prepared as in Examples 1-3 was added and mixed under a nitrogen blanket, and the mixture 5 was again heated to and held at 70°C. An IR spectrum taken after the mixture had been held at 70°C for six hours exhibited no isocyanate band (2270 cm⁻¹).

The identities and quantities of polyether polyols, diisocyanates, and secondary amine functional alkoxy silanes used and the quantities of Dabco T-12 used are shown in Table 2. The viscosities and theoretical molecular weights of the 10 resulting alkoxy silane functional poly(ether-urethane)s are shown in Table 3. Note that Examples C10-C15 are comparative.

Table 2

Ex.	Polyether Polyol	Polyether Polyol wt (mcq OH)	Diisocyanate	Diisocyanate wt (mcq NCO)	Silane wt (meq OH)	Dabco T-12 wt
4	Voranol 222-029 and Vorano 235-056	180 g (92.1) and 20 g (20.0)	Desmodur I	18.88 g (168.1)	23.97 g (56.0) of Example 1	0.20 g
5	Pluronic L62	200 g (160)	Desmodur I	25.04 g (223)	26.96 g (63) of Example 1	0.40 g
6	Pluronic L92	200 g (110)	Desmodur I	19.20 g (171)	26.26 g (61) of Example 1	0.40 g
7	Pluronic 17R2	200 g (186)	Desmodur I	28.06 g (250)	27.32 g (64) of Example 1	0.40 g
8	Pluronic L92	667.95 g (366)	Desmodur I	63.67 g (567)	90.25 g (201) of Example 2	1.34 g
9	Multranol 9111	200 g (100)	Desmodur I	18.09 g (161)	26.13 g (61) of Example 1	0.40 g
C10	Pluronic L61	200 g (200)	Desmodur I	29.68 g (264)	27.52 g (64) of Example 1	0.40 g
C11	Pluronic L101	200 g (105)	Desmodur I	18.70 g (166)	26.20 g (61) of Example 1	0.40 g
C12	Pluronic 31R1	200 g (123)	Desmodur I	20.76 g (185)	26.45 g (62) of Example 1	0.40 g
C13	Arcol E351	200 g (143)	Desmodur I	23.04 g (205)	26.72 g (62) of Example 1	0.40 g
C14	Pluronic L92	200 g (110)	Mondur TD-80	14.80 g (170)	25.92 g (61) of Example 1	0.10 g
C15	Pluronic L92	164.25 g (90)	Desmodur I	20.21 g (180)	37.58 g (90) of Example 3	0.33 g

Table 3

Example	Alkoxy silane Functional Poly(ether-urethane) Viscosity	Alkoxy silane Functional Poly(ether-urethane) Molecular Weight
4	81.0 Pa sec	8700
5	69.3 Pa sec	8000
6	42.5 Pa sec	8000
7	149.0 Pa sec	8000
8	28.5 Pa sec	8200
9	68.5 Pa sec	8000
C10	129.5 Pa sec	8000
C11	29.8 Pa sec	8000
C12	61.0 Pa sec	8000
C13	84.8 Pa sec	8000
C14	185.2 Pa sec	8000
C15	not measured	4900

Procedure for the Preparation of Moisture Curable Sealant Compositions

Moisture curable sealant compositions were prepared from each of the
 5 alkoxy silane terminated poly(ether-urethanes) prepared in Examples 4-9 and
 Comparative Examples C10-C15 according to the generalized procedure below.
 The actual component weights used to prepare each moisture curable sealant
 composition are shown in Table 4.

A vacuum reactor fitted with a high shear mixer was flushed with nitrogen
 10 and filled with the following components:

200 grams alkoxy silane functional poly(ether-urethane) of Examples 4-9 and
 Comparative Examples C10-C15 (as specified in Table 5),
 Jayflex DIDP (as specified in Table 4),
 2 grams Tinuvin 770,
 15 8 grams Irganox 1010, and
 0.1 grams Elftex 8.

The components were mixed at low shear for 5 minutes while under a
 nitrogen blanket. Dislon 6500 (see Table 4) was added and mixing occurred at high
 sheer for 10 minutes. The particulate fillers (selected from Ultrapflex, Pfinyl 402,

and Zeeospheres 600 as specified in Table 4, previously dried overnight at 105°C) were added and mixed at high speed under full vacuum for 30 minutes while heating the mixture to 90°C. The mixture was then cooled to 50°C, and the reactor purged with nitrogen to break the vacuum. A solution containing Isopar H, Dabco T-12, 5 and Silquest A171 in the amounts specified in Table 4 was added and mixed at high shear for 1 hour under nitrogen at 50°C. A solution containing Silquest A1120, Neostann U220, and N-methylpyrrolidinone (NMP) in the amounts specified in Table 4 was then added and mixed at high shear for 15 minutes under nitrogen at 10 50°C. A vacuum was applied to the resulting mixture to remove all nitrogen bubbles. The mixture was then immediately transferred into thick walled high density polyethylene cartridges.

Table 4

	Example					
	16	17	18	19	20	21
Jayflex DIDP wt.	100 g	100 g	100 g	100 g	100 g	100 g
Dislon 6500 wt.	10 g	10 g	10 g	10 g	8 g	10 g
Ultrapflex wt.	50 g	50 g	50 g	50 g	200 g	50 g
Pfinyl 402 wt.	525 g	550 g	550 g	550 g	400 g	550 g
Zeeospheres 600 wt.	0	0	0	0	0	0
Isopar H wt.	24 g	30 g	30 g	30 g	34 g	50 g
Silquest A171 wt.	6 g	6 g	6 g	6 g	6 g	6 g
Dabco T-12 wt.	0.1 g	0.1 g	0.1 g	0.1 g	0.2 g	0.2 g
Silquest A1120 wt.	12 g	10 g	10 g	10 g	10 g	10 g
NMP wt.	26 g	20 g	20 g	20 g	24 g	0
Neostann U220 wt.	2 g	2 g	2 g	2 g	4 g	3 g

Table 4 (continued)

	Comparative Examples					
	C22	C23	C24	C25	C26	C27
Jayflex DIDP wt.	100 g	100 g	100 g	100 g	100 g	120 g
Dislon 6500 wt.	10 g	10 g	10 g	10 g	8 g	12 g
Ultrapflex wt.	50 g	50 g	50 g	50 g	200 g	0
Pfinyl 402 wt.	550 g	550 g	550 g	550 g	400 g	400 g
Zeeospheres 600 wt.	0	0	0	0	0	250 g
Isopar H wt.	30 g	30 g	30 g	50 g	48 g	30 g
Silquest A171 wt.	6 g	6 g	6 g	6 g	6 g	6 g
Dabco T-12 wt.	0.1 g	0.1 g	0.1 g	0.2 g	0.2 g	0.2 g
Silquest A1120 wt.	10 g	10 g	10 g	10 g	10 g	12 g
NMP wt.	20 g	20 g	20 g	0	0	12 g
Neostann U220 wt.	2 g	2 g	2 g	3 g	2 g	4 g

The moisture curable sealant compositions of Examples 16-21, C22-C27 performed acceptably when tested in accordance with the tack free time, hardness, adhesion, cold flexibility, wet-on-wet paintability, and shelf life tests described above. Results of the viscosity and paint adhesion tests are shown in Table 4.

Table 5

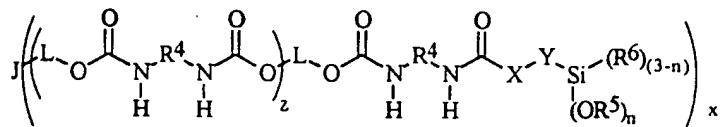
Ex.	Alkoxysilane Functional Poly(ether- urethane)	Paint adhesion ASTM 3359B		
		1 hour	24 hour	72 hour
16	Example 4	5B	5B	3B
17	Example 5	2B*	5B	4B
18	Example 6	5B	5B	4B
19	Example 7	4B	5B	5B
20	Example 8	5B	4B	4B
21	Example 9	5B	4B	4B
C22	Example C10	5B	5B	0B
C23	Example C11	1B	2B	1B
C24	Example C12	2B	2B	1B
C25	Example C13	4B	3B	3B
C26	Example C14	4B	0B	0B
C27	Example C15	3B	2B	3B

* denotes cohesive failure within the cured moisture curable sealant composition, not adhesive failure at the interface of paint and cured moisture curable sealant composition.

5 It will be apparent to those skilled in the art that various modifications and variations can be made in the method and article of the present invention without departing from the spirit or scope of the invention. Thus, it is intended that the present invention cover the modifications and variations of this invention provided they come within the scope of the appended claims and their equivalents.

What is claimed is:

1. A moisture curable sealant composition comprising an alkoxy silane functional poly(ether-urethane) having formula (I):
 5 (I)



wherein:

J is selected from the group consisting of alkyl, aryl, alkylene, alkarylene, and arylene with valence x and a number average molecular weight of less than 10 about 250;

each L is independently selected from the group consisting of (i) a polyether segment comprising ethylene oxide (EO) units distributed randomly or in blocks, the segment comprising greater than about 15 mol % and less than about 40 mol % EO and the segment having a number average molecular weight of about 2000 to 15 about 8,000, and (ii) a divalent alkylene group having a number average molecular weight of less than about 200;

each R⁴ is independently selected from the group consisting of C₂ to C₁₀ alkylene and cycloalkylene;

each z is independently selected from integers greater than or equal to zero;
 20 each X is independently selected from the group consisting of O, S, and NR⁷, wherein R⁷ is selected from the group consisting of hydrogen, aryl, alkyl, -CH₂CH(R)W¹ and -CH(W²)CH₂W¹, wherein

each R is independently selected from the group consisting of C₁ to C₄ alkyl; and

25 W¹ and W² are independently selected from the group consisting of -CO₂R⁹, -CONR⁹₂, -CN, and -COR⁹; or W¹ and W² taken together form a divalent linking group selected from the group consisting of -CO₂-R⁸-O₂C-, -CONR⁹-R⁸-NR⁹CO-, -CONR⁹-R⁸-O₂C-, and -CONR⁹CO- such that W¹, W², and an ethylene group through which they are joined form a 5 to 8 member ring, wherein each R⁸ is 30 independently selected from the group consisting of alkylene having from 2 to 10

carbon atoms; each R⁹ is independently selected from the group consisting of hydrogen, alkyl and aryl;

each Y is independently selected from the group consisting of linear, branched, and cyclic alkylene groups having at least two carbon atoms;

5 each R⁵ is independently selected from the group consisting of alkyl, alkenyl, aryl, acyl, and N-ketiminy;

each R⁶ is independently selected from the group consisting of hydrogen, alkyl, and aryl;

each n is independently selected from integers 1 to 3; and

10 each x is independently selected from integers greater than or equal to 1.

2. The composition of claim 1 wherein each L is independently selected from the group consisting of (i) a polyether segment comprising an internal polypropylene oxide block and terminal polyethylene oxide blocks, the segment comprising 20 to 30 mol % EO and the segment having a number average molecular weight of about 3000 to about 6000.

15

3. The composition of claim 1 wherein each R⁴ is independently selected from a group consisting of 1,6-hexylene and isophorone-diy.

20

4. The composition of claim 1 which has at one hour after coating a paint adhesion of at least 4B as measured by ASTM 3359B.

5. The composition of claim 1 which has at 24 hours after coating a paint adhesion of at least 4B as measured by ASTM 3359B.

25

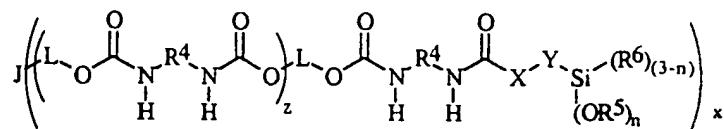
6. The composition of claim 1 which has at 72 hours after coating a paint adhesion of at least 3B as measured by ASTM 3359B.

30

7. A moisture curable sealant composition comprising a homogenous blend comprising:

(a) 100 parts by weight of an alkoxy silane functional poly(ether-urethane) having formula (I):

(I)



5

wherein:

J is selected from the group consisting of alkyl, aryl, alkylene, alkarylene, and arylene with valence x and a number average molecular weight of less than about 250;

10

each L is independently selected from the group consisting of (i) a polyether segment comprising ethylene oxide (EO) units distributed randomly or in blocks, the segment comprising greater than about 15 mol % and less than about 40 mol % EO and the segment having a number average molecular weight of about 2000 to about 8,000, and (ii) a divalent alkylene group having a number average molecular weight of less than about 200;

15

each R^4 is independently selected from the group consisting of C_2 to C_{10} alkylene and cycloalkylene;

each z is independently selected from integers greater than or equal to zero;

20

each X is independently selected from the group consisting of O, S, and

NR^7 , wherein R^7 is selected from the group consisting of hydrogen, aryl, alkyl, $-\text{CH}_2\text{CH}(\text{R})\text{W}^1$ and $-\text{CH}(\text{W}^2)\text{CH}_2\text{W}^1$, wherein

each R is independently selected from the group consisting of C_1 to C_4 alkyl; and

25

W^1 and W^2 are independently selected from the group consisting of $-\text{CO}_2\text{R}^9$,

$-\text{CONR}^9$, $-\text{CN}$, and $-\text{COR}^9$; or W^1 and W^2 taken together form a divalent linking group selected from the group consisting of $-\text{CO}_2\text{R}^8-\text{O}_2\text{C}-$, $-\text{CONR}^9\text{R}^8-\text{NR}^9\text{CO}-$, $-\text{CONR}^9\text{R}^8-\text{O}_2\text{C}-$, and $-\text{CONR}^9\text{CO}-$ such that W^1 , W^2 , and an ethylene group through which they are joined form a 5 to 8 member ring, wherein each R^8 is independently selected from the group consisting of alkylene having from 2 to 10

carbon atoms; each R^y is independently selected from the group consisting of hydrogen, alkyl, and aryl;

5 each Y is independently selected from the group consisting of linear, branched, and cyclic alkylene groups having at least two carbon atoms;

 each R^5 is independently selected from the group consisting of alkyl, alkenyl, aryl, acyl, and N-ketiminy;

 each R^6 is independently selected from the group consisting of hydrogen, alkyl, and aryl;

10 each n is independently selected from integers 1 to 3; and
 each x is independently selected from integers greater than or equal to 1;

 (b) 10 to 100 parts by weight of at least one plasticizer;

 (c) 0.1 to 10 parts by weight of at least one antioxidant;

15 (d) 1 to 5 parts by weight of at least one catalyst;

 (e) 0.1 to 10 parts by weight of at least one adhesion promoter;

 (f) 0 to 10 parts by weight of at least one dehydrator;

20 (g) 0 to 400 parts by weight of at least one filler;

 (h) 0 to 10 parts by weight of at least one rheology modifier; and
 (i) 0 to 40 parts by weight of an organic solvent.

25

8. The composition of claim 11 wherein each L is independently selected from the group consisting of (i) a polyether segment comprising an internal polypropylene oxide block and terminal polyethylene oxide blocks, the segment comprising 20 to 30 mol % EO and the segment having a number average molecular weight of about 3000 to about 6000.

30

9. The composition of claim 11 wherein each R^4 is independently selected from the group consisting of 1,6-hexylene and isophorone-diyl.

35

10. An article comprising a laminate, wherein the laminate comprises:

5 (a) a substrate;

(b) a cured layer of the moisture curable sealant composition of claim 11 bonded to at least a portion of the substrate; and

(c) a layer of one or more of the following coated over the layer of the moisture curable sealant composition: (i) paint, (ii) primer, (iii) paint sealer.

10

INTERNATIONAL SEARCH REPORT

In. International Application No

PCT/US 97/19367

A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 C08G18/10 C09J171/02 C08G65/32 C08G18/48

According to International Patent Classification(IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C08G C09J C09D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0 410 199 A (BAYER AG) 30 January 1991 see example 2	1-6
Y	---	7-10
Y	EP 0 459 304 A (BASF CORP) 4 December 1991 see claim 1 see page 3, line 24 - line 25 see page 4, line 16 - line 58	7-10
X	EP 0 269 819 A (BAYER AG) 8 June 1988 see example 2	1-6
X	US 5 364 955 A (ZWIENER CHRISTIAN ET AL) 15 November 1994 cited in the application see example 8	1-6

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Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

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INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 97/19367

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 210 444 A (UNION CARBIDE CORP) 4 February 1987 see claim 1 see page 6, line 9 - line 17 ---	1-10
A	EP 0 164 520 A (HENKEL KGAA) 18 December 1985 see claims 1,4,7 see page 8, line 3 - line 9 -----	1-10

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Information on patent family members

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